



SECTION 1. IDENTIFICATION

Product Identifier Sulfur (Solid or Molten)

Other Means of Identification

Sulphur (Solid), Sulphur (Molten), Brimstone

Product Family Petroleum Residue

Recommended Use Raw material for the manufacture of sulphuric acid, fertilizers, rubber vulcanization, and other

chemical processes.

Restrictions on UseNot recommended for uses other than those listed, or for non-industrial purposes.

Manufacturer/Supplier

Keyera and Affiliates

Identifier

Suite 200, The Ampersand, West Tower

144 - 4th Avenue SW Calgary, Alberta T2P 3N4

Main Phone No. (403) 205-8300 / 1 (888) 699-4853 (Mon. - Fri. 8 AM - 5 PM)

Transportation CANUTEC (CAN), Ph.: 1-888-CAN-UTEC (226-8832) Cell: *666, (24 hr)

Emergencies Only CHEMTREC (US), 1-800-424-9300, (24 hr)

SECTION 2. HAZARD IDENTIFICATION

Classification

Flammable liquid - Category 1; Combustible dust - Category 1; Skin corrosion - Category 1

Label Elements







Signal Word: Danger

Hazard Statement(s):

Flammable solid.

Combustible liquid.

Causes severe skin burns and eye damage. May ignite and burn at elevated temperatures.

May be corrosive to metals.

Harmful if inhaled.

Precautionary Statement(s):

Do not handle until all safety precautions have been read and understood.

Avoid breathing dust/fume/gas/mist/vapours/spray.

Do not get in eyes, on skin, or on clothing.

Wash hands and skin thoroughly after handling.

Do not eat, drink or smoke when using this product.

Wear protective Sulfur (Solid or Molten)

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Response:

IF SWALLOWED: Immediately call a POISON CENTRE or doctor.

IF ON SKIN: Immerse in cool water or wrap in wet bandages.

Take off contaminated clothing and wash it before reuse.

IF INHALED: If breathing is difficult, remove person to fresh air and keep comfortable for breathing.

IF INHALED: Call a POISON CENTRE or doctor if you feel unwell.

IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present

and easy to do. Continue rinsing.

Storage:

Store in a well-ventilated place. Keep container tightly closed.

Store in accordance with local, regional, national and international regulations.

Disposal:

Dispose of contents and container in accordance with local, regional, national and international regulations.

Other Hazards

Molten material may contain dangerous levels of hydrogen sulfide and sulfur dioxide.

Contact with hot product will cause thermal burns.

CONTAINS HYDROGEN SULFIDE. Product may contain significant quantities of dissolved hydrogen sulfide gas. H2S has a broad range of effects dependent on the airborne concentration and length of exposure: 0.02 ppm odor threshold, smell of rotten eggs; 10 ppm eye and respiratory tract irritation; 100 ppm coughing, headache, dizziness, nausea, eye irritation, loss of sense of smell in minutes; 200 ppm potential for pulmonary edema after >20-30 minutes; 500 ppm loss of consciousness after short exposures, potential for respiratory arrest; >1000 ppm immediate loss of consciousness, may lead rapidly to death, prompt cardiopulmonary resuscitation may be required. Do not depend on sense of smell for warning. H2S causes rapid olfactory fatigue (deadens sense of smell). There is no evidence that H2S will accumulate in the body tissue after repeated exposure.

| SECTION 3. COMPOSITION/INFORMATION ON INGREDIENTS | | | | |
|---|-----------|---------|---------------------------|--|
| Chemical Name | CAS No. | % | Other Identifiers | |
| Sulfur (Solid or Molten) | 7704-34-9 | 99.9+ | Brimstone | |
| Hydrogen Sulfide | 7783-06-4 | 0 -0.1 | Sulphur hydride, acid gas | |
| Sulfur dioxide | 7446-09-5 | 0 - 0.1 | Not available | |

Notes

Concentrations are expressed in % weight/weight.

May contain traces of hydrogen sulfide and/or sulfur dioxide.

SECTION 4. FIRST-AID MEASURES

First-aid Measures

Inhalation

CONTAINS HYDROGEN SULFIDE. In case of oxygen deficiency: take precautions to ensure your own safety before attempting a rescue (e.g. wear appropriate protective equipment). Move to fresh air. Do NOT perform rescue breathing if the victim inhaled or ingested the material; induce artificial respiration with a respiratory medical device. Obtain medical attention immediately.

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Skin Contact

Take off immediately contaminated clothing, shoes and leather goods (e.g. watchbands, belts). Wash off immediately with plenty of water for at least 15 minutes. If persistent irritation occurs, obtain medical attention. If in contact with molten material, flush area with lukewarm water immediately. Do NOT attempt to pull off clothing or solidified material from the skin! Obtain medical attention immediately.

Eye Contact

If irritation or redness develops from exposure, flush eyes with clean water. If persistent irritation occurs, obtain medical attention.

For contact with hot material:

Gently open eyelids and flush affected eye(s) with cold (not icy) water. Seek medical attention immediately.

For contact with cold material:

Immediately flush the contaminated eye(s) with lukewarm, gently flowing water for 15-20 minutes, while holding the eyelid(s) open. If eye irritation persists, get medical advice or attention.

Ingestion

Not a likely route of exposure. Rinse mouth with water. Obtain medical attention.

Most Important Symptoms and Effects, Acute and Delayed

CONTAINS HYDROGEN SULPHIDE. Hydrogen sulfide is extremely toxic. H2S has a broad range of effects dependent on the airborne concentration and length of exposure: 0.02 ppm odour threshold, smell of rotten eggs; 10 ppm eye and respiratory tract irritation; 100 ppm coughing, headache, dizziness, nausea, eye irritation, loss of sense of smell in minutes, 200 ppm potential for pulmonary edema after >20-30 minutes; 500 ppm loss of consciousness after short exposures, potential for respiratory arrest; >1000 ppm immediate loss of consciousness, may lead rapidly to death, prompt cardiopulmonary resuscitation may be required. Do NOT depend on sense of smell for warning. H2S causes rapid olfactory fatigue (deadens sense of smell). There is no evidence that H2S will accumulate in the body tissue after repeated exposure.

Contact with molten material:

Will burn the skin. Permanent scarring may result.

Immediate Medical Attention and Special Treatment

Special Instructions

Treat symptomatically. Hydrogen sulfide (H2S) - CNS asphyxiant. May cause rhinitis, bronchitis, and occasionally pulmonary edema after severe exposure. Consider oxygen therapy. Consult a Poison Control Centre for guidance.

SECTION 5. FIRE-FIGHTING MEASURES

Extinguishing Media

Suitable Extinguishing Media

Carbon dioxide, dry chemical powder, appropriate foam, water spray or fog. Sand (smothering agent).

Unsuitable Extinguishing Media

None known.

Specific Hazards Arising from the Product

Molten material can cause severe thermal burns upon contact. CONTAINS HYDROGEN SULFIDE: TOXIC BY INHALATION.

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Special Protective Equipment and Precautions for Fire-fighters

Fight fire from a safe distance or a protected location. For a massive fire, immediately evacuate the area and use unmanned hose holder or monitor nozzles. Wear full protective clothing and self-contained breathing apparatus. Chemical protective clothing (e.g. chemical splash suit) and positive pressure SCBA may be necessary.

SECTION 6. ACCIDENTAL RELEASE MEASURES

Personal Precautions, Protective Equipment, and Emergency Procedures

Evacuate the area immediately. Isolate the hazard area. Keep out unnecessary and unprotected personnel. Use the personal protective equipment recommended in Section 8 of this safety data sheet. Eliminate all ignition sources. Use grounded, explosion-proof equipment. Remove or isolate incompatible materials as well as other hazardous materials.

Environmental Precautions

It is good practice to prevent releases into the environment. Do not allow into any sewer, on the ground or into any waterway. If the spill is inside a building, prevent product from entering drains, ventilation systems and confined areas.

Methods and Materials for Containment and Cleaning Up

Stop or reduce leak if safe to do so. Do not use absorbents. Contain spill using noncombustible material such as vermiculite, earth or sand. Do NOT use combustible materials such as sawdust. Place used absorbent into suitable, covered, labelled containers for disposal. Avoid generating dust. Avoid dry sweeping. If necessary, use a dust suppressant such as water. Do not use compressed air for clean-up. Use water fog or spray curtain to reduce amount of dust in air.

SECTION 7. HANDLING AND STORAGE

Precautions for Safe Handling

Solid Sulfur does not represent significant hazards. Avoid generating dusts. Eliminate heat and ignition sources such as sparks, open flames, hot surfaces and static discharge. Post "No Smoking" signs. Do not use near welding operations or other high energy sources. Do not weld, cut or perform hot work on empty container until all traces of product have been removed. Good housekeeping is extremely important. Prevent dust accumulation on ALL surfaces including ceiling rafters and other hidden surfaces. Prevent accidental contact with incompatible chemicals. Adequate ventilation is required when working in areas containing or when handling molten Sulfur. Molten Sulfur should not be put into any tank, rail car, or truck trailer that contains trace quantities of hydrocarbons, or more than trace amounts of moisture. This material may contain or release poisonous hydrogen sulfide gas.

Thoroughly clean clothing, shoes and leather goods before reuse or dispose of safely. Launder clothes before rewearing. Inform laundry personnel of product hazard(s). Properly dispose of any contaminated items, including shoes, that cannot be decontaminated. DO NOT re-use.

Conditions for Safe Storage

Molten Sulfur storage and loading/unloading temperature range is 127°C to 150°C (260°F to 300°F). Store in a well ventilated area away from all sources of ignition. Avoid storage in confined spaces or near incompatible materials, oxidizers, or materials that support combustion. Protect from conditions listed in Conditions to Avoid in Section 10 (Stability and Reactivity).

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| SECTION 8. EXPOSURE CONTROLS/PERSONAL PROTECTION | | | | | | | |
|--|-----------|------------|----------|----------|----------|-----------|--|
| Control Parameters | ACGIH TLV | ACGIH TLV® | | OSHA PEL | | AIHA WEEL | |
| Chemical Name | TWA | STEL | TWA | Ceiling | 8-hr TWA | TWA | |
| Sulfur (Solid or Molten) | | | 10 mg/m3 | | | | |
| Hydrogen Sulfide | 1 ppm | 5 ppm | | 20 ppm | | | |
| Sulfur dioxide | 2 ppm | 5 ppm | | | | | |

Appropriate Engineering Controls

Do not allow product to accumulate in the air in work or storage areas, or in confined spaces. Use local exhaust ventilation, if general ventilation is not adequate to control amount in the air. If current ventilation practices are not adequate to maintain airborne concentrations below the established exposure limits, additional engineering controls may be required.

Individual Protection Measures

Eye/Face Protection

When working with molten Sulfur:

Wear chemical safety goggles and face shield when contact is possible.

Skin Protection

The use of insulated gloves impervious to the material handled is advised to prevent skin contact. Users should check with manufacturers to confirm the breakthrough performance of their products. Depending on exposure and use conditions, additional protection may be necessary to prevent skin contact, including the use of items such as chemical resistant boots, aprons, arm covers, hoods, coveralls, or encapsulated suits.

Respiratory Protection

Where there is potential for airborne exposure to hydrogen sulfide (H2S) above exposure limits, a NIOSH approved self-contained breathing apparatus (SCBA) or equivalent, operated in a pressure demand or other positive pressure mode should be used. Under conditions where hydrogen sulfide is not detected, a NIOSH certified air purifying respirator equipped with R or P95 filters may be used.

SECTION 9. PHYSICAL AND CHEMICAL PROPERTIES

Basic Physical and Chemical Properties

Appearance Yellow solid.

Odour Slight hydrocarbon, possibly faint rotten egg smell

Odour Threshold 0.005 - 0.01 ppm (Hydrogen Sulfide)

Melting Point/Freezing Point 121.7 °C (melting); Not available (freezing)

Boiling point/Initial boiling point 444.6 °C

Flash Point 188 °C (closed cup)

Evaporation Rate Not applicable
Flammability (solid, gas) Flammable solid.

Upper/Lower Flammability or Not available (upper); Not available (lower)

Explosive Limit

Vapour Pressure Not available
Vapour Density (air = 1) Not applicable

Relative Density (water = 1) 1.80 at 120°C (as liquid sulphur)

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Solubility Insoluble in water; Soluble in aromatic hydrocarbons (e.g. toluene).

Partition Coefficient, n- Not available

Octanol/Water (Log Kow)

Auto-ignition Temperature 190 °C

Decomposition Temperature Not available

Viscosity Not available (kinematic)

Other Information

Physical State Solid Molecular Formula S, S8

Molecular Weight 32 g/mol as S, 256.5 as S8

Surface Tension 60.8 dynes/cm at 120°C (as liquid sulphur)

SECTION 10. STABILITY AND REACTIVITY

Reactivity

Not reactive under normal conditions of use.

Chemical Stability

Unstable under certain conditions - see Conditions to Avoid.

Possibility of Hazardous Reactions

Contact with incompatible materials may cause fire, excessive heat generation, uncontrolled reaction, release of toxic products and/or explosion.

Conditions to Avoid

High temperatures. Open flames, sparks, static discharge, heat and other ignition sources. Generation of dust. Prolonged contact with water, moisture or humidity.

Incompatible Materials

Reacts with: chlorates, nitrates, oxidizing agents (e.g. peroxides), carbides, halogens (e.g. chlorine), phosphorus, heavy metals.

Corrosive to: carbon steel, iron, steel, and other metals.

Hazardous Decomposition Products

Sulfur burns to sulfur dioxide. Sulfur reactions with hydrocarbons and other organic materials may produce hydrogen sulfide and carbon disulfide. Other possible toxic reactions or decomposition products are highly dependent on the incompatible material.

SECTION 11. TOXICOLOGICAL INFORMATION

Likely Routes of Exposure

Inhalation; skin contact; eye contact; ingestion.

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Acute Toxicity

| Sulfur (Solid or Molten) | >9.23 mg/L (rat) (4-hour | >3000 mg/kg (rat) | >2000 mg/kg (rabbit) |
|--------------------------|--------------------------|------------------------|----------------------|
| | exposure) | | |
| Hydrogen Sulfide | 444 ppm (rat) (4-hour | Not available | Not available |
| | exposure) | | |
| Sulfur dioxide | | 2520 ppm (rat) (1-hour | Not available |
| | | exposure) | |

Skin Corrosion/Irritation

Human experience shows mild irritation. Elevated temperature material can cause severe burns.

Serious Eye Damage/Irritation

Human experience shows mild irritation. May be irritating to eyes. Symptoms may include redness, swelling, pain, tearing, and blurred or hazy vision.

STOT (Specific Target Organ Toxicity) - Single Exposure

Inhalation

May cause nose and throat irritation.

Skin Absorption

Hot material can cause severe burns.

Ingestion

Large doses (15 grams) by mouth may lead to hydrogen sulphide production in the body, chiefly due to bacterial action within the colon.

Aspiration Hazard

Not known to be an aspiration hazard.

STOT (Specific Target Organ Toxicity) - Repeated Exposure

Prolonged inhalation of dust over several years may cause respiratory complications such as asthma and inflammation of the frontal and maxillary sinuses, and may cause respiratory diseases like emphysema and bronchiectasis.

Respiratory and/or Skin Sensitization

Not known to be a respiratory sensitizer. Not known to be a skin sensitizer.

Carcinogenicity

Not specifically listed.

Reproductive Toxicity

Development of Offspring

Not hazardous according to OSHA/WHMIS criteria.

Sexual Function and Fertility

Not hazardous according to OSHA/WHMIS criteria.

Effects on or via Lactation

No information was located.

Germ Cell Mutagenicity

Not hazardous according to OSHA/WHMIS criteria.

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Interactive Effects

None known.

SECTION 12. ECOLOGICAL INFORMATION

Ecotoxicity

No information was located.

Persistence and Degradability

No ingredient of this product or its degradation products is known to be highly persistent.

Bioaccumulative Potential

This product and its degradation products are not likely to bioaccumulate.

Mobility in Soil

If released into the environment, this product is not expected to move through the soil, based on physical and chemical properties.

Other Adverse Effects

There is no information available.

SECTION 13. DISPOSAL CONSIDERATIONS

Disposal Methods

Disposal should be in accordance with applicable regional, national, and local laws and regulations. Local regulations may be more stringent than regional or national requirements and must be complied with.

SECTION 14. TRANSPORT INFORMATION

| Regulation | UN No. | 11. 2 11. 3 | Transport Hazard Class(es) | Packing Group |
|--------------|--------|--------------------|----------------------------|------------------|
| Canadian TDG | 1350 | SULPHUR, or SULFUR | 4.1 | III |
| US DOT | 1350 | SULPHUR, or SULFUR | 4.1 | III |

| Canadian TDG 2 | | MOLTEN SULFUR; MOLTEN SULPHUR; SULFUR, MOLTEN: or SULPHUR, MOLTEN | 4.1 | III |
|----------------|------|---|-----|-----|
| US DOT 2 | 2448 | MOLTEN SULFUR; MOLTEN SULPHUR; SULFUR, MOLTEN; or SULPHUR, MOLTEN | 4.1 | III |

Environmental

Hazards

Environmentally Hazardous Substance

Special Precautions

Please note: CONTAINS HYDROGEN SULFIDE : May evolve hazardous levels of H2S in

headspace volumes of closed tanks and vessels. TOXIC BY INHALATION.

TDG Schedule 2, Special Provision 32 & 33 may apply, depending on road & rail transport container markings and/or the products particular shape it has been formed to (e.g. Prills, granules, flakes, etc.).

want in Bully according to Intermedianal Maritime Consultation In

Transport in Bulk according to International Maritime Organization Instruments: Not applicable

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SECTION 15. REGULATORY INFORMATION

Safety, Health and Environmental Regulations

This section is not required by WHMIS.

SECTION 16. OTHER INFORMATION

NFPA Rating Health - Not assigned. Flammability - 1 Instability - 0

SDS Prepared By

Bureau Veritas Canada

Phone No.

1-800-386-7247

Date of Preparation
Date of Last Revision

August 05, 2021 August 05, 2021

Revision Indicators

All sections revised form original Keyera SDS last revision date of August 31, 2015.

Key to Abbreviations

ACGIH® = American Conference of Governmental Industrial Hygienists

OSHA = US Occupational Safety and Health Administration RTECS® = Registry of Toxic Effects of Chemical Substances

References

CHEMINFO database. Canadian Centre for Occupational Health and Safety (CCOHS).

Registry of Toxic Effects of Chemical Substances (RTECS®) database. Dassault

Systèmes/BIOVIA ("BIOVIA"). Available from Canadian Centre for Occupational Health and

Safety (CCOHS).

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information related to an unusual or other use.

SDS representative location(s):

Rimbey Facility
Simonette Facility

Cheecham (Sulphur Handling) Facility